

THE STEREOSPECIFIC CYCLOADDITION OF CHLOROSULFONYL
ISOCYANATE TO CIS- AND TRANS- β -METHYLSTYRENE AND CIS-
AND TRANS-3-HEXENE

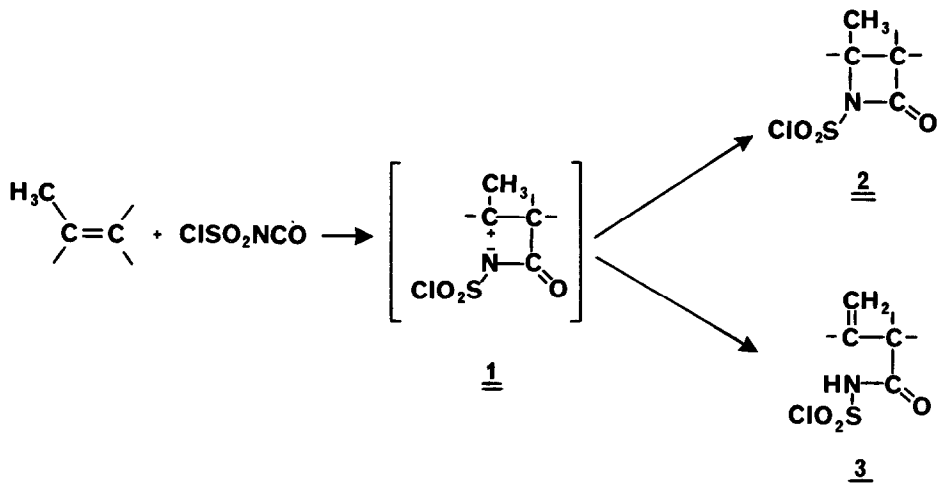
Emil J. Moriconi and John F. Kelly

Department of Chemistry, Fordham University

New York, N. Y.

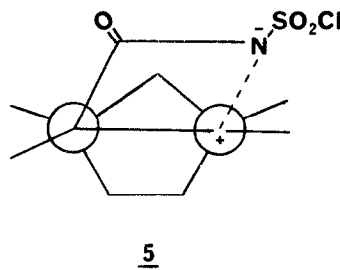
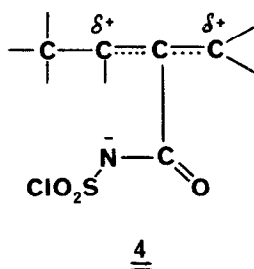
(Received in USA 28 November 1967)

In the cycloaddition of chlorosulfonyl isocyanate (CSI) to olefins, Graf^{1,2} envisioned a two-step mechanism involving the initial formation of a 'dipolar adduct' 1 which could stabilize itself through closure to the four-membered 2 and/or via proton shift to 3. Similarly, we suggested for the reaction between CSI and allenes an allyl-type stabilized carbonium ion 4 in the transition state.³ Most recently we reported the cycloaddition of CSI to bridged



bicyclic (norbornene, norbornadiene and bicyclo[2.2.2]octene) and tricyclic (endo- and exo-dicyclopentadiene) olefins leading, in each case, to a single, unrearranged N-chlorosulfonyl- β -lactam cycloadduct.⁴ Since a thermal 2+2 concerted cycloaddition is not allowed by the

Woodward-Hoffman selection rules,⁵ we suggested a pseudo-concerted reaction⁶ involving the intermediacy of a 1,4-dipole (5) in which the charged species are aligned for bonding.⁴

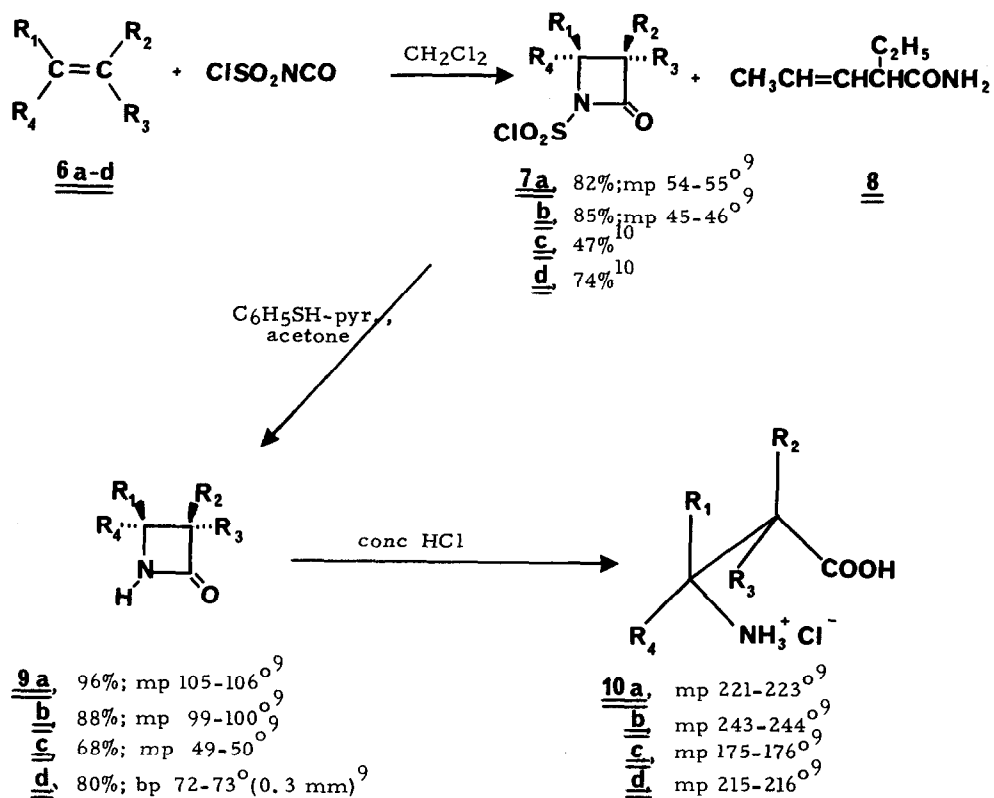


We now wish to report the stereospecific cis-addition of CSI to cis- (6a) and trans- β -methylstyrene (6b) to yield 2+2 cycloadducts, N-chlorosulfonyl-cis- (7a)⁹ and trans-3-methyl-4-phenyl-2-azetidinone (7b),⁹ respectively (Scheme I). Similarly, cis-(6c) and trans-3-hexene (6d) led to N-chlorosulfonyl-cis-(7a)¹⁰ and trans-3,4-diethyl-azetidinone (7d),¹⁰ in addition to small amounts (< 5%) of 2-ethyl-3-pentenamide (8).⁹ The retention of configuration of R₁-R₄ in 7a-d is unequivocally supported by nmr data. Thus the eclipsed (dihedral $\angle \approx 0^\circ$) cis-protons in 7a and 7c show the expected vic-coupling of 7.25-7.50 cps while the trans-skewed protons (dihedral $\angle \approx 120^\circ$) in 7b and 7d display vic-coupling of 3.75-4.00 cps. Further, the methyl protons (R₂) in 7a are in the shielding region of the cis-phenyl ring (R₁) and appear as a doublet upfield (0.56 ppm) relative to the trans-methyl protons (R₃) in 7b.¹¹

N-Chlorosulfonyl- β -lactams 7a-d were reduced with benzenethiol and pyridine in acetone^{1,2} to the appropriate unsubstituted β -lactams, 9a-d. Concentrated hydrochloric acid hydrolysis quantitatively converted 9a and 9b to erythro- (10a)⁹ and threo-3-amino-2-methyl-3-phenylpropanoic acid hydrochloride (10b),⁹ respectively, while meso- (10c) and dl-3-amino-2-ethylpentanoic acid hydrochloride (10d)⁹ were similarly obtained from 9c and 9d.

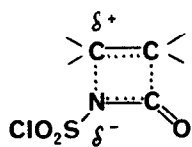
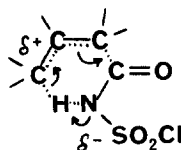
The stereoselective cis-addition of CSI to the title compounds eliminate, at least, the intermediacy of 1 enroute to 7a-d. Three mechanisms now seem available to account for the

SCHEME I



- a, R₁=C₆H₅; R₂=CH₃; R₃=R₄=H
b, R₁=C₆H₅; R₃=CH₃; R₂=R₄=H
c, R₁=R₂=C₂H₅; R₃=R₄=H
d, R₁=R₃=C₂H₅; R₂=R₄=H

stereospecificity observed: (i) a pseudo-concerted reaction leading to 5; here one must assume either that the collapse of the internal ion-pair to β -lactam product occurs so rapidly as to preclude rotational isomerism, or that such rotation is hindered by dipolar interaction; (ii) a near-concerted process described by 11; ¹² and (iii) a concerted cycloaddition in which case the Woodward-Hoffman rules for thermal 2+2 cycloadditions must be modified for such cumulenic

1112

systems such as CSI perhaps as a consequence of d-orbital participation of the S-atom and/or the lone pair of electrons on the N-atom. ¹³ Mechanisms i-iii each is probably preceded by a π -complex. ¹⁴ Unsaturated amide 8 could be formed via 1 either directly or by the thermally reversible ring-openings of 2 \rightleftharpoons 1 \rightarrow 3, ¹⁵ and/or by a concerted addition-elimination involving a cyclic transition state 12.

REFERENCES

- 1) R. Graf, Ber., 89, 1071 (1956); Org. Syntheses, 46, 23 (1966).
- 2) R. Graf, Ann., 661, 111 (1963); Org. Syntheses, 46, 51 (1966).
- 3) E. J. Moriconi and J. F. Kelly, J. Am. Chem. Soc., 88, 3657 (1966).
- 4) E. J. Moriconi and W. C. Crawford, J. Org. Chem., 33, in press (1968).
- 5) R. Hoffman and R. B. Woodward, J. Am. Chem. Soc., 87, 395, 2046 (1965)
- 6) J. C. Martin, P. Glenn Gott, V. Wilson Goodlett and R. H. Hasek, J. Org. Chem., 30, 4175 (1965).
- 7) R. Huisgen, Angew. Chem. Intern. Ed., 2, 633 (1963); R. Huisgen, R. Grashey and J. Sauer in "The Chemistry of Alkenes," S. Patai, Ed., Interscience Publishers, New York, N. Y., 1964, p. 739.
- 8) S. Proskow, H. E. Simmons and T. L. Cairns, J. Am. Chem. Soc., 88, 5254 (1966).

- 9) These compounds were analyzed for C, H and N and acceptable analyses ($\pm 0.3\%$) were obtained.
- 10) These compounds decomposed on analyses using vpc and combustion techniques. Nmr spectra, however, indicated at least 95% purity.
- 11) K. D. Barrow and T. M. Spotswood, Tetrahedron Letters, 3326 (1965).
- 12) W. T. Brady and H. R. O'Neal, J. Org. Chem., 32, 612, 2704 (1967).
- 13) Suggested by Prof. R. Huisgen, Private communication.
- 14) Tetraphenylethylene, e.g., gave an intense blue-violet colored solution upon addition of CSI in methylene chloride. Addition of water to this solution led to its vigorous reaction with CSI, immediate loss of color and recovery of the olefin.
- 15) F. Effenberger and R. Gleiter, Ber., 97, 1576 (1964).

Acknowledgement. This research was supported by Public Health Service Research Grant No. 1-RO1-A108-063-01 from the National Institute of Allergy and Infectious Diseases.